

DOTT.SSA ANNA PAPPALARDO
UDIENZA 19.06.2026
AULA 2 CAMILLERI-PIAZZA VERGA PIANO TERRA

Udienze predibattimentali:

- 1) 09:00 ~~Andriano Giovanni~~ 7344/25
- 2) 09:00 ~~Tiralongo Paolo~~ 6744/25
- 3) 09:00 ~~Ramoni Davide~~ 7132/25
- 4) 09:00 1) ~~Pulvrenti Salvatore~~, 2) ~~Maria Orsini~~, 3) ~~Luigi Menegazzi~~: (2016/24)
- 5) 09:10 ~~Rizzo Sebastiano Antonino~~: 2884/26
- 6) 09:10 ~~Romano Massimo~~: 2898/26
- 7) 09:10 1) ~~Sgari Stefano~~, 2) ~~Ferraro Sergio Ugo Luciano~~: 2885/26
- 8) 09:10 ~~Delucchi Giovanni~~: 2846/26
- 9) 09:15 1) ~~Mancuso Daniela~~, 2) ~~Loguano Guido~~, 3) ~~Cuffanti Hilda~~, 4) ~~Trisanti Ruggia~~, 5) ~~Tomaselli Ignazio Giovanni~~: 7523/25
- 10) 09:15 ~~Garofa Bogdan Ionel~~: 2881/26
- 11) 09:15 ~~Ciliberto Giacomo~~: 2882/26

Prime udienze:

- 12) 09:15 Coco Concetto: (art.385) per scegliere il rito. 3705/24
- 13) 09:20 1) ~~Michael Kevin Edward~~, 2) ~~Funes Norma Ibeth~~, 3) ~~Farrar Tiara Lyvette~~, 4) ~~Blanco Keth Lasha~~: 2949-2026
- 14) 09:20 ~~Amis Camilla~~: 2964/26
- 15) 09:30 Maione ~~Emanuele~~

Processi in cui verrà svolta attività (istruttoria e/o discussione)

- 16) 09:30 Pastura ~~Andrea~~
- 17) 10:00 Bonaccorsi ~~Andrea~~
- 18) 10:00 ~~Battista Angelo Salvatore~~: 4219/22

THE UNIVERSITY OF CHICAGO
DEPARTMENT OF CHEMISTRY
PHYSICAL CHEMISTRY LABORATORY

1. The first part of the experiment is devoted to the study of the temperature dependence of the rate constant for the reaction of hydrogen peroxide with iodide ions in the presence of ceric ions as a catalyst. The reaction is carried out in a series of solutions of known concentrations of the reactants and the products are measured at various temperatures. The rate constants are determined from the initial rates of the reaction and the Arrhenius plot is constructed from the logarithm of the rate constants versus the reciprocal of the absolute temperature. The activation energy of the reaction is determined from the slope of the straight line obtained.

2. The second part of the experiment is devoted to the study of the effect of the concentration of the reactants on the rate of the reaction. The reaction is carried out in a series of solutions of known concentrations of the reactants and the products are measured at various concentrations. The rate constants are determined from the initial rates of the reaction and the order of the reaction with respect to each reactant is determined from the log-log plot of the rate versus the concentration of the reactant. The overall order of the reaction is also determined.

- 19) 10:00 ~~Ventura Luca~~: 6628/17
- 20) 10:00 ~~Amadori Giuliano~~: 1673/23
- 21) 10:00 ~~Cantarella Mario - Grasso Salvatore~~: 1860/25
3158/20
- 22) 10:00 ~~Jovani Saverio~~ in videocollegamento
- 23) 10:30 ~~Musumeci Vincenzo~~: 4777/17
- 24) 10:30 ~~Fichera Fabio Orazio~~: 914/19
- 25) 11:00 ~~Totolo Beniamino~~: 4730/19
- 26) 11:00 ~~Bracciolone Alessio Giovanni~~: 4628/17
- 27) 11:00 ~~Sapienza Roberta~~: 7608/23
- 28) 11:30 ~~Ballato Antonio Concetto~~: 5719/25
- 29) 11:30 ~~Tommaselli Mario~~: 903/25
- 30) 11:30 ~~Catalano Vincenzo~~: 2177/24
- 31) 11:30 ~~Tommaselli Mario~~: ~~5885/16~~, 903/25
- 32) 12:00 ~~Mirabella Luciano~~: 5885/16
- 33) 12:00 ~~Bonichi Dario~~: 5877/22
- 34) 12:30 ~~Di Masi Masquero John Finer~~: 5874/25
- 35) 13:00 ~~Fuglisi Francesco~~: 7133/25
- 36) 13:00 ~~Capini Salvatore~~: 4864/19
- 37) 13:00 ~~Viglianesi Sebastiano Marcello~~: 5594/25
- 38) 13:30 ~~Ignaccolo Michela~~: 7443/23
- 39) 13:30 ~~Nicolesi Carmela~~: 6482/19
- 40) 13:30 ~~Condorelli Giuseppe~~: 6199/22
- 41) 13:30 ~~Napoli Naomi~~: 3745/22
- 42) 15:00 ~~Patanè Orazio~~
- 43) 16:00 ~~Carona Domenico~~: 5893/24

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